
Investigating the Influence of pH on Drug Release from Zidovudine Matrices Containing Different Grades of Hydroxypropyl Methyl Cellulose

Masheer Ahmed Khan^{1*}

DOI: 10.9734/bpi/cacs/v6/1927C

ABSTRACT

The current study investigates the impact of a multimedia dissolution profile on the drug release of zidovudine-containing sustained-release hydrophilic matrices using a variety of hydroxypropyl methylcellulose grades. Zidovudine, the first anti-HIV drug to receive FDA approval for clinical use, is widely used to treat AIDS, either alone or in combination with other antiviral drugs. To sustain the release of the medication, matrices were created utilising a combination of several grades of HPMC, namely HPMCK4M and HPMCK15M. In-vitro multimedia dissolution studies were carried out in order to mimic the in-vivo situation. The pH/buffer is chosen based on the drug's exposure from the stomach to the intestine/colon. The research ensures that pH changes have an effect on drug dissolution and release for absorption. Matrices also provide quite regulated release of the drug zidovudine over an extended period of time.

Keywords: Multimedia; dissolution; matrices; pH.

1. INTRODUCTION

Zidovudine, the first anti-HIV drug to receive FDA approval for clinical use, is widely used to treat AIDS, either alone or in combination with other antiviral drugs. Its dose-dependent haematological toxicity, low therapeutic index, short biological half-life, and poor bioavailability are the key limitations to its therapeutic effectiveness. It is rapidly absorbed from the gastrointestinal tract (GIT) after oral administration, reaching a peak plasma concentration of 1.2 µg/mL at 0.8 hours. It is first transformed to zidovudine triphosphate in the systemic circulation, which is pharmacologically active and suppresses HIV viral replication. Since zidovudine -triphosphate has a biological half-life of 4 hours, it must be administered often (3 to 4 times per day) to maintain therapeutic medication levels. Since zidovudine acts as a metabolic antagonist of thymidine and its antiviral effect is time dependent, an adequate zero-order delivery of zidovudine is desired for maintaining anti-AIDS effect and avoiding the strong side effects. These side effects are usually associated with excessive plasma level of zidovudine immediately after intravenous or oral administration. Zidovudine is absorbed throughout the GIT. The drug is freely soluble at any pH and hence judicious selection of release retarding excipients is necessary for achieving constant in vivo release. The most commonly used method of modulating the drug release is to include it in a matrix system [1-6].

Multimedia dissolution is to mimic the in-vivo condition by doing in-vitro test and pH/buffer selection is based on the exposure of drug from stomach to intestine/colon and to ensure the impact of pH changes on dissolution and release of drug substance for absorption [7-9]. In vitro dissolution is well established as a quality control technique to monitor the batch-to-batch quality and performance of a drug product. Dissolution is routinely used by most pharmaceutical firms to guide the development of new formulations and to monitor product quality after scale-up of batch size, and after changes in formulation, manufacturing process, equipment, and site of manufacture [10,11].

¹School of Pharmacy, Devi Ahilya Vishwavidyalaya, Takshshila Campus, Khandwa Road, Indore, 452001, India.

*Corresponding author: E-mail: masheerak@yahoo.com;

Sustained release drug delivery system of zidovudine is designed to achieve a prolonged therapeutic effect by continuously releasing medication over an extended period of time by using different grades of Hydroxypropyl methylcellulose (HPMC) viz. HPMCK4M and HPMCK15M [12-16].

2. MATERIALS AND METHODS

Zidovudine was obtained as a gift sample and tablets were prepared by direct compression using HPMCK4M and HPMCK15M polymer combinations. Other excipients used were Magnesium stearate, Talc, MCC and dibasic calcium phosphate. The drug was analyzed by UV spectrophotometer (UV 1601 Shimadzu, Japan) at 266 nm.

3. RESULTS AND DISCUSSION

Physical properties of the tablets were found within the probable limits as shown in Table 1. The drug content was estimated from the absorbance obtained. Three tablets of zidovudine were placed into three different pH of phosphate buffer (pH2.4, pH 6.8 and pH 7.4). The USP dissolution apparatus was set at rotation 50 rpm and temperature of the assembly was set at 37°C. Absorbance was measured at 266 nm by collecting sample at different time intervals up to 12 hrs. The percentage drug release was calculated at different time intervals at different pH and shown in Table 2. The graph was plotted between percent drug release and time for different dissolution media and shown in Fig. 1.

Table 1. Physical characteristics of the tablets

FORMULATION				
HPMC K4M mg	HPMC K15M mg	Weight mg Mean ± SD	Hardness Kg Mean ± SD	Friability (%)
25	15	120 ± 1.96	5.50 ± 0.12	0.50-0.08

Table 2. Result of dissolution studies with different pH

Sr. No.	Time hrs.	Absorbance (nm.)			% Drug release		
		2.4	6.8	7.4	pH2.4	pH6.8	pH7.4
1	0.5	0.161	0.168	0.222	12.78	13.5	20.3
2	1	0.186	0.194	0.438	16.16	16.53	26.4
3	2	0.214	0.224	0.573	19.68	22.24	38.5
4	4	0.321	0.361	0.692	31.69	36.24	55.6
5	6	0.431	0.582	0.732	45.22	61.2	71.2
6	8	0.446	0.612	0.811	46.27	65.52	78.5
7	12	0.567	0.68	0.904	60.97	84.1	95.5

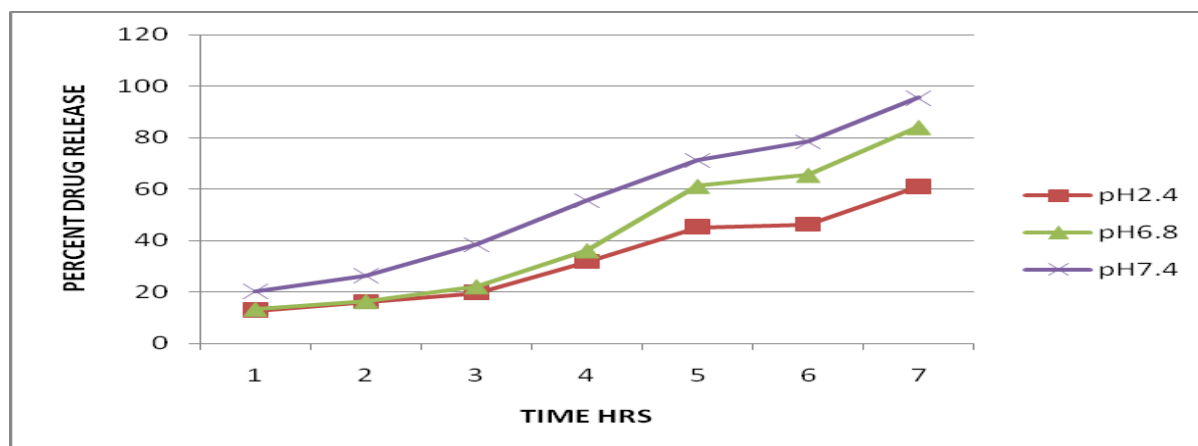


Fig. 1. Percent drug release V/s Time in different pH media

4. CONCLUSION

The release profile of zidovudine from the matrices increased continuously with time, and the amount of drug release best seen in acidic media (pH=7.4). The cumulative amount of drug release is higher at pH 7.4 than that of pH 6.8 by 11.4 % and then that of pH 2.4 by 24.53 %. This increase in drug release at higher pH can be attributed to pH dependent solubility of stavudine. As the pH increases, the solubility of zidovudine increases which might increase drug release from matrices.

COMPETING INTERESTS

Author has declared that no competing interests exist.

REFERENCES

1. Goodman and Gilman's. The Pharmacological basis of therapeutics. 10th edition. Mc-Graw Hill. 2001;709-710.
2. Ansel HC, Aallen LV, Popovich NG. Pharmaceutical dosage forms and drug delivery system" Lippincott Williams & Wilkins, Philadelphia (USA). 2002;234-235.
3. Kuksal A, Tiwary KA, Jain NK, Jain S. Formulation and in vitro, in vivo evaluation of extended-release matrix tablet of Zidovudine: Influence of combination of hydrophilic and hydrophobic matrix formers. AAPS Pharm Sci Tech. 2006;7(1):E1-E9.
4. Ganesh S, Radhakrishnan M, Ravi M, Prasannakumar B, Kalyani J. In vitro Evaluation of the Effect of Combination of Hydrophilic and Hydrophobic Polymers on Controlled Release Zidovudine Matrix Tablets. Indian J Pharm Sci. 2008;70(4):461-5.
5. Salsa T, Veiga F, Pina ME. Oral controlled-release dosage forms. I. Cellulose ether polymers in hydrophilic matrices. Drug Dev Ind Pharm. 1997;23:929-938.
6. Huang P, Farquhar D, Plunkett W. Selective action of 2',3' dideoxy-2',3'-dideoxythymidinetriphosphate on human immune deficiency virus reverse transcriptase and human DNA polymerases. J. Biol. Chem. 1982;267:1817-1822.
7. Khan MA. Studies on diltiazem hydrochloride sustained release matrices profiles in multimedia dissolution conditions, Research Journal of Pharmaceutical, Biological and Chemical Sciences. 2013;4(2):1573-1576. ISSN:0975-8585.
8. Khan MA, Mehta RK. Studies on Multimedia dissolution Profile of Zolpidem Tartrate Sustained Release Matrix Tablets, Research Journal of Pharmaceutical, Biological and Chemical Sciences (RJPBCS). 2012;3(2):174-177. ISSN:0975-8585.
9. Khan MA. Studies on the effect of pH over dissolution profile of diclofenac sodium sustained release tablets. Journal of Drug Delivery and Therapeutics. 2012;2(5):65-66.
10. Shah VP, Lesko LJ. Current challenges and future regulatory directions in *in vitro* dissolution. Drug Information Journal. 1995;29(3):885-91.
11. Anand OM, Lawrence XY, Conner DP, Davit BM. Dissolution testing for generic drugs: an FDA perspective. The AAPS Journal. 2011;13(3):328-35. Liam C Feely, Stanley S Davis. The influence of polymeric excipients on drug release from hydroxyl propyl methyl cellulose matrices. International Journal of Pharmaceutics. 1988;44:131-139.
12. Khan MA, Chaturvedi SC. Swelling and Drug Release Studies from Hydrophilic Matrices Containing Combination of Different Grades of Hydroxyl Propyl Methylcellulose. Asian Journal of Chemistry. 2011;23(8):3566–3568.
13. Khan MA. Studies of swelling effect and drug release in hydrophilic matrices containing different grades of polymers, Research J. of Pharm. Biological and Chemical Sci. 2013;4(1):1241-1247.
14. Khan MA, Maheshwari RK. Studies of relationship between swelling and drug release in the sustained release hydrophilic matrices containing different grades of hydroxyl propyl methylcellulose. Research J. of Pharm. Biological and Chemical Sci. 2011;2(4):970-975.
15. Swarbric J, Boylan JC. Encyclopedia of Pharmaceutical Technology, Marcell Dekker, New York; 1995.

Biography of author(s)



Masheer Ahmed Khan

School of Pharmacy, Devi Ahilya Vishwavidyalaya, Takshshila Campus, Khandwa Road, Indore, 452001, India.

He is an Assistant Professor at Devi Ahilya University, Indore, MP, India with an extensive teaching experience of more than 18 years. Formerly into the field of Pharma Marketing, he always had a passion for research and teaching. Industrial Pharmacy is the specialization of author and the thrust area of research is formulation and optimization of various dosage forms. With a sound research background, he has published around 40 research articles in various reputed national and international journals. He has worked as a Principal Investigator on Major research projects, obtained from University Grants Commission (UGC), New Delhi, and Madhya Pradesh Council of Science and Technology (MPCOST), Bhopal, MP. He is a Supervisor and guide to students pursuing Ph.D. He has attended many pharmaceutical conferences, seminars and workshops, refresher courses and orientation programs. He is a Life member of various Professional bodies, like IPA, APTI etc. He has an active participation in corporate life with different capacities.

© Copyright (2021): Author(s). The licensee is the publisher (B P International).

DISCLAIMER

This chapter is an extended version of the article published by the same author(s) in the following journal. Asian Journal of Pharmaceutical Science & Technology, 4(1), 1-3, 2014.